

NEW DEGRADED QUINONE DITERPENOID FROM THE STEMS OF *Byrsonima coccolobifolia* Kunth. (Malpighiaceae)

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A chemical investigation of two specimens of *Byrsonima coccolobifolia* collected in the southeast cerrado and from central Brazil was performed. A new degraded diterpenoid, byrsonimaquinone, was isolated from the stems along with known compounds. This is the first study on the roots of *B. coccolobifolia*, and several triterpenes, such as α -amyrin, β -amyrin, oleanolic acid, and glochidonol, along with a mixture of stigmasterol, β -sitosterol and campesterol, were identified. These compounds were identified by spectroscopic analysis techniques, including 1D and 2D NMR, GC-MS and high-resolution mass spectrometry.

Keywords: Malpighiaceae; *Byrsonima coccolobifolia*; degraded diterpenoid.

INTRODUCTION

The genus *Byrsonima*, a member of the Malpighiaceae family, contains 150 species and is widespread in tropical America. In Brazil, species of the genus *Byrsonima* are known as “muricis” and can be found in north, northeast, southeast and central regions of the cerrado (savannah).¹ The cerrado is the second largest Brazilian biome and a global hotspot of biodiversity, which makes further understanding the chemistry of these organisms quite valuable in view of increasing deforestation.²

Byrsonima coccolobifolia Kunth. is a medicinal plant used to treat stomach disorders and as an antidiarrheal.^{1,3} Previously, the extracts of the leaves have been shown to have antibacterial and molluscicidal activities and the ability to inhibit arginase, a molecular target of *Leishmania amazonensis*.³⁻⁵

The compounds most commonly isolated from this genus are flavonoids, terpenes, gallic acids and quinic acid derivatives.¹ Among the constituents, flavonoids (flavanones, bioflavonoids, flavonols and procyanidins), triterpenes (with oleanolic and ursolic skeletons) and modified triterpenoids (steroids) are commonly found in *B. basiloba*,^{6,7} *B. bucidifolia*,⁸ *B. crassa*,^{9,10} *B. crassifolia*,¹¹⁻¹⁴ *B. fagifolia*,¹⁵⁻¹⁷ *B. intermedia*,¹⁸ *B. microphylla*,^{19,20} and *B. verbascifolia*.^{21,22}

Previous studies on the leaves of *B. coccolobifolia* revealed the presence of flavonoids, gallic acid and xanthones.^{23,24} Recently, we investigated the ethyl acetate extracts of the leaves and stems of this species, and guided by arginase inhibition, we isolated flavonoids and tannins with potent inhibitory activities against arginase (IC₅₀ values ranging from 0.13 to 4.8 $\mu\text{mol L}^{-1}$).^{4,5}

In our on-going studies of the chemistry of *B. coccolobifolia*, we have investigated two specimens collected in the southeast (stems and leaves) and central (roots) cerrado of Brazil. As part of our analysis of the sample collected in the central cerrado, from its roots, we only isolated triterpenes and steroids such as α -amyrin, β -amyrin, oleanolic acid and glochidonol (Figure 1). Our investigation of its leaves and stems led to the isolation of a new diterpene derivative,

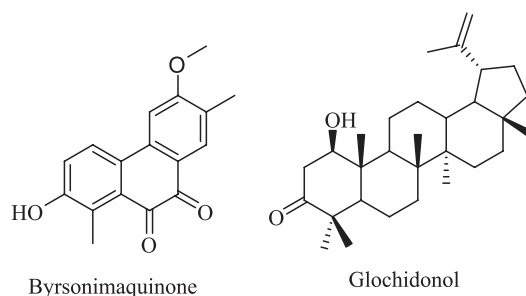


Figure 1. *Byrsonimaquinone*, a new degraded quinone diterpenoid isolated from *B. coccolobifolia* stems. *Glochidonol*, a triterpenoid described for the first time in a member of the *Byrsonima* genus, was isolated from *B. coccolobifolia* roots

byrsonimaquinone, in addition to other known compounds that we have described before.^{4,5}

EXPERIMENTAL SECTION

General experimental procedures

¹H-NMR and 2D NMR spectra were obtained on a Bruker model ARX 100 MHz and DRX-400 NMR spectrometer. Me₄Si was used as an internal standard (*J* in Hz), and CDCl₃, CD₃OD and DMSO-d₆ were used as solvents. MS spectra (*m/z*) were acquired using a Bruker Daltonics, Micromass TOF - Q II - ESI - TOF, high-resolution mass spectrometer.

For GC-MS analysis, an Agilent GC-7820A model gas chromatograph – mass spectrometer coupled to an MSD 5975 was used. Experimental conditions were as follows: 10 μL of sample was injected in the GC-MS in the splitless mode; starting at 120 °C (2 min), then increasing 10 °C/min until 250 °C, then increasing at 2 °C/min until 275 °C, and then 35 °C/min until 310 °C (10 min); the split was 20:1; the flow rate was 1 mL/min; the column was an HP-5 (30 m x 250 μm and film 0.25 μm ; Agilent); and the data were collected in the electron impact ionization mode at 70 eV.

For chromatography, silica gel (SiO₂) (Merck, 60-200 mesh; and Macherey-Nagel, 70-230 and 230-400 mesh) and Sephadex LH-20 (Amersham Pharmacia Biotech AB) were used. The solvents were

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ethanol (EtOH), methanol (MeOH), hexane, dichloromethane, and ethyl acetate (EtOAc), and they were purchased from Vetec. Thin-layer chromatography (TLC) was carried out with F_{254} , $\phi = 0.2$ mm, pre-coated aluminum silica 60 (20 x 20 cm) (Whatman, Fluka and Merck). TLC plates were visualized using UV_{254/366} light as well as a sulfuric vanillin solution.

Plant material

B. coccolobifolia materials were collected in distinct regions. Stems were collected from the cerrado at the Federal University of São Carlos (UFSCar), São Carlos - SP, Brazil, in July 2011. This species was identified by Dr. Maria Inês Salgueiro Lima and deposited at the Herbarium of the Botany Laboratory (HUFSCar) at UFSCar (Voucher n° 8367).

The roots were collected in central Brazil in the Federal District - Gama - DF, Brazil, in December 2010. The sample of *B. coccolobifolia* was identified by Dr. Helder Nagai Consolaro and deposited at the EMBRAPA Herbarium - Genetic Resources and Biotechnology (CEN) (Voucher n° BW 6029).

Extraction and isolation

The crude extract of the roots (65 g) of *B. coccolobifolia* was obtained after percolating the dried powder three times in ethanol for six days each time. Ethanolic extracts from the stems (30.5 g) were obtained from dried powder macerated with EtOH at room temperature over two weeks as described previously.^{4,5} Both extracts were concentrated by rotatory evaporation at 40 °C. The crude extract from the roots was suspended in a mixture of MeOH/H₂O (3:7), and the crude extract from the stems was suspended in EtOH/H₂O (1:3). Both suspensions were sequentially partitioned in hexane and EtOAc. From the roots, this procedure generated 2.5 g of material from the hexane layer (BcRH), 24.1 g of material from the EtOAc layer (BcRA), and 38.5 g of material from the water-alcoholic layer (BcRW). The liquid-liquid partitioning of the crude stem extract led to 1.3 g of material from the hexane layer (BcSH), 10.0 g of material from the EtOAc layer (BcSA), and 19.0 g of material from the water-alcoholic layer (BcSW).

Compounds isolation from stem extracts

The EtOAc extract from the stems, BcSA (10.0 g), was chromatographed on a column with a height (h) of 12.0 cm and a diameter (ϕ) of 5.0 cm (SiO₂, 60-200 mesh; isocratic mode with hexane/acetone (9:1)) and yielded four fractions. Fraction 4 (4.0 g) was chromatographed several times as follows: Sephadex LH-20, h = 7.0 cm, $\phi = 5.0$ cm, isocratic mode with MeOH; fraction four from the Sephadex column (55.0 mg) was further purified over silica gel (SiO₂, 230-400 mesh, h = 26.0 cm, $\phi = 2.0$ cm, isocratic mode with CH₂Cl₂/MeOH using 1% of MeOH); fraction fourteen (15 mg) from the silica column was fractionated over Sephadex LH-20, h = 54.0 cm, $\phi = 1.6$ cm, isocratic mode with MeOH; and fraction 8 from that column contained the diterpene byrsonimaquinone (1.8 mg). MS and 1D and 2D NMR were performed, and the results were compared to the literature data.²⁵

Byrsonimaquinone. TOF - Q II - ESI - TOF - MS: [2 M + Na]⁺ *m/z*: 587.1678; [M + Na]⁺ *m/z*: 305.0780 and M of *m/z*: 282.0898 (C₁₇H₁₄O₄, calcd 282.0892); ¹H and 2D NMR (Table 1).

The compounds 5-*O*-galloylquinic acid, 3,5-di-*O*-galloylquinic acid, 5-*O*-(3-methylgalloyl)-quinic acid, 3,4,5-tri-*O*-galloylquinic acid, gallic acid, (+)-catechin, (-)-epicatechin, quercitrin, isoquercitrin, (+)-syringaresinol and trigonostemone were isolated from the leaves and stems and were characterized in our previous work.^{4,5}

Compounds isolated from the root extracts

BcRH (2.5 g) was subjected to column chromatography (CC) (SiO₂, 70-230 mesh; h = 11.0 cm, $\phi = 5.5$ cm; gradient mode with hexane/EtOAc and EtOAc/MeOH). The second fraction (1.6 g) obtained from this column was repurified by CC (SiO₂, 70-230 mesh; h = 12.0 cm, $\phi = 5.0$ cm; gradient mode with hexane/EtOAc and EtOAc/MeOH) and afforded five fractions after TLC analysis (F₂₋₁, F₂₋₂, F₂₋₃, F₂₋₄ and F₂₋₅), and the second of those fractions (0.8 g) was a mixture of α - and β -amyrin.

The third fraction, F₂₋₃ (0.1 g), was further purified using CC (SiO₂, 230-400 mesh; h = 23.0 cm, $\phi = 2.5$ cm; gradient mode with hexane/EtOAc and EtOAc/MeOH). From the four fractions afforded by that column (F₃₋₁, F₃₋₂, F₃₋₃, F₃₋₄), F₃₋₂ (0.1 g) was further purified using CC (SiO₂, 230-400 mesh; h = 30.0 cm, $\phi = 1.7$ cm; gradient mode with hexane/CH₂Cl₂ and CH₂Cl₂/EtOAc), which lead to the isolation of glochidonol (3.0 mg). Fractions F₄₋₄ and F₄₋₅ contained a mixture of phytosterols (5.0 mg), namely, stigmasterol, β -sitosterol and campesterol.

The EtOAc extract BcRA (24.1 g) was fractionated using CC with SiO₂, and after seven chromatography steps, oleanolic acid (9.7 mg) was obtained. Briefly, the CC conditions for the isolation of oleanolic acid are as follows: the first CC of BcRA (24.1 g) was SiO₂, 70-230 mesh, h = 15.0 cm, $\phi = 5.5$ cm, gradient mode with hexane/EtOAc and EtOAc/MeOH; fraction two from first column (4.6 g) was used for the second column, which was SiO₂, 70-230 mesh, h = 9.0 cm, $\phi = 5.5$ cm, gradient mode with hexane/EtOAc and EtOAc/MeOH; fraction one from the second column (4.6 g) was subjected to a third column using SiO₂, 70-230 mesh, h = 12.0 cm, $\phi = 5.5$ cm, gradient mode with hexane/EtOAc and EtOAc/MeOH; fraction one from the third CC (1.7 g) was subjected to a fourth CC using SiO₂, 230-400 mesh, h = 24.5 cm, $\phi = 2.5$ cm, gradient mode with hexane/EtOAc and EtOAc/MeOH; fraction three from the fourth column (0.2 g) was fractionated again over SiO₂, 230-400 mesh, h = 21.0 cm, $\phi = 2.5$ cm, gradient mode with hexane/EtOAc and EtOAc/MeOH; fraction two of the fifth column (53.8 mg) was purified again using SiO₂, 230-400 mesh, h = 26.0 cm, $\phi = 1.7$ cm, gradient mode with hexane/EtOAc and EtOAc/MeOH; and fraction two from the sixth column (18 mg) was fractionated over SiO₂, 230-400 mesh, h = 21.0 cm, $\phi = 1.5$ cm, gradient mode with hexane/EtOAc and EtOAc/MeOH. Fraction two of the seventh column contained oleanolic acid. The compounds were identified by ¹H-NMR and 2D NMR and GC/MS, and the results were compared to the literature data.²⁶⁻³⁰

RESULTS AND DISCUSSION

The EtOAc extracts of *B. coccolobifolia* stems and leaves were fractionated, and the compounds isolated include 5-*O*-galloylquinic acid, 3,5-di-*O*-galloylquinic acid, 5-*O*-(3-methylgalloyl)-quinic acid, 3,4,5-tri-*O*-galloylquinic acid, gallic acid, (+)-catechin, (-)-epicatechin, quercitrin, isoquercitrin, (+)-syringaresinol and trigonostemone.^{4,5} A new degraded diterpenoid purified from stems was named byrsonimaquinone (Figure 1).

In a second study, the hexane and EtOAc extracts of the roots of the specimen collected in the central Brazilian cerrado were submitted to several chromatographic columns, which eventually lead to the identification of the triterpenes α -amyrin, β -amyrin,²⁶ oleanolic acid,²⁷ glochidonol,²⁸ and the mixture of stigmasterol, β -sitosterol and campesterol.^{29,30}

The new quinone diterpene derivative was identified by ¹H NMR and 2D NMR experiments, such as HSQC, HMBC, and 1D NOE (Table 1), as well high-resolution mass spectrometry. The mass spectrum was obtained in the positive mode showing an ion signal

at m/z 587.1678 corresponding to the sodiated dimer, $[2M + Na]^+$, of the diterpene. The base peak, observed at m/z 305.0780, can be attributed to the sodiated diterpene molecular ion, which indicates a molecular formula of $C_{17}H_{14}O_4$ (calc. 282.0892).

In the 1H NMR spectrum of the diterpene, there are signals from aromatic hydrogens at δ_H 8.03 (d ; $J = 9.0$ Hz) and 7.17 (d ; $J = 9.0$ Hz), indicating an *ortho* relationship between hydrogens H-1 and H-2. H-1 is more deshielded due to its *meta* relationship to the hydroxyl group instead of the *ortho* relationship observed in manniorthoquinone.²⁵ The chemical shifts of the singlets at δ_H 7.69 and 7.47 indicate hydrogens H-14 and H-11 on the second aromatic ring moiety are *para* to each other.

The proton at δ_H 4.01 (s) was attributed to the methoxyl group bound to an aromatic ring and the two singlets at δ_H 2.41 and 2.16 (18-CH₃ and 15-CH₃) are characteristic of methyl groups bound to aromatic rings, and they are bound to C-4 and C-13, respectively.

Table 1. 1H (400 MHz) and ^{13}C NMR (100 MHz) signals of byrsonimaquinone. Data were acquired from heteronuclear correlation and 1D NOE analysis in DMSO- d_6 . Chemical shifts are reported in δ (ppm) and coupling constants (J) are reported in Hz

C	HSQC		HMBC	1D NOE
	δ_C	δ_H	$^2J_{HC}; ^3J_{HC}; ^4J_{HC}$	
1	123.5	8.03 (d , 9.0)	H-11	H-11
2	120.2	7.17 (d , 9.0)		H-1
3	156.9		H-1, 18-CH ₃	
4	129.3		H-2	
5	130.7		H-1, 18-CH ₃	
6	—			
7	179.8		H-14	
8	—			
9	137.5		H-1, H-11, H-14	
10	127.4		H-2, H-11	
11	104.8	7.47 (s)		H-1
12	163.8		H-11, 12-OCH ₃ , H-14, 15-CH ₃	
13	126.6		15-CH ₃	
14	131.2	7.69 (s)	15-CH ₃	15-CH ₃
15-CH ₃	15.3	2.16 (s)	H-14	
18-CH ₃	13.2	2.41 (s)		
12-OCH ₃	56.0	4.01 (s)		H-11

The HSQC indicated that the aromatic hydrogens at δ_H 8.03 (d ; $J = 9.0$ Hz) and 7.17 (d ; $J = 9.0$ Hz) are directly correlated ($^1J_{HC}$) with C-1 (δ_C 123.5 ppm) and C-2 (δ_C 120.2 ppm), respectively. The hydrogens on the second aromatic ring, δ_H 7.69 (s) and 7.47 (s), are coupled to C-14 (δ_C 131.2 ppm) and C-11 (δ_C 104.8), respectively. The methyl group hydrogens of 18-CH₃ correlate with C-18 (δ_C 13.2), and the hydrogens of 15-CH₃ correlate with C-15 (δ_C 15.3). The hydrogens of the 12-OCH₃ methoxyl group at δ_H 4.01 (s) correlate with the carbon at δ_C 56.0 ppm.

In comparing the data obtained here for isolated byrsonimaquinone with those reported in the literature for manniorthoquinone,²⁵ the differences were primarily related to the positioning of the substituents around the aromatic ring as seen in the different chemical shifts, multiplicities and coupling constants in the NMR spectra. Hydroxyl groups are inductively withdrawing and therefore deshield

C-3 (δ_C 156.9). However, hydroxyl groups are electron donating by resonance and therefore shield C-4 (δ_C 129.3). In byrsonimaquinone, C-5 is *meta* to the hydroxyl group, so it is more deshielded (δ_C 130.7) than in manniorthoquinone²⁵ where it is *para* to the hydroxyl group.

The heteronuclear long-range couplings and 1D NOE experiments together with all the NMR and MS data allowed us to propose the molecular structure of the new degraded diterpenoid (Figure 2). The substitution arrangement on the aromatic rings was established based on HMBC correlations (Figure 2). The $^3J_{HC}$ of H-14 (δ_H 7.69) with C-9 (δ_C 137.5), C-12 (δ_C 163.8), 15-CH₃ (δ_C 15.3) and C-7 (δ_C 179.8) provided evidence that the methoxyl group was attached to this ring and that the carbonyl is next to H-14. In addition, the hydrogens of the methoxyl group (δ_H 4.01) correlate to C-12 (δ_C 163.8), and the hydrogens of the 15-CH₃ methyl (δ_H 2.16) correlate with C-12 (δ_C 163.8), C-13 (δ_C 126.6) and C-14 (δ_C 131.2).

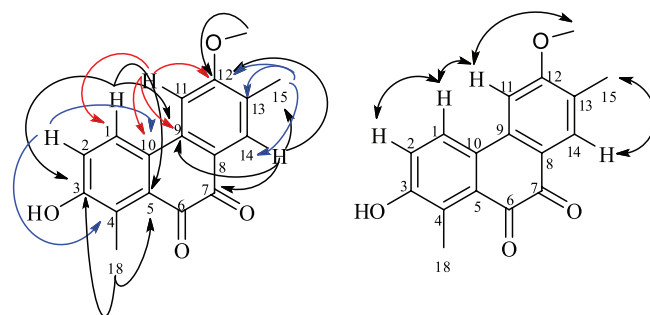


Figure 2. Heteronuclear correlations. Long-range couplings in the HMBC spectrum of byrsonimaquinone: \curvearrowright ; 1D NOE interactions of 1: \curvearrowright

The $^2J_{HC}$ cross-peaks between H-11 (δ_H 7.47) with C-9 (δ_C 137.5) and C-12 (δ_C 163.8), as well as the $^3J_{HC}$ of H-11 to C-10 (δ_C 127.4) are in agreement with those reported for manniorthoquinone.²⁵ A $^4J_{HC}$ coupling between H-11 (δ_H 7.47) and C-1 (δ_C 123.5) was also observed.

For the second aromatic ring, the spectrum showed $^3J_{HC}$ correlations of H-1 (δ_H 8.03) with C-3 (δ_C 156.9), C-5 (δ_C 130.7) and C-9 (δ_C 137.5), which provides evidence that the hydroxyl group is attached at the more deshielded carbon (C-3). H-2 (δ_H 7.17) correlates to C-4 (δ_C 129.3) and C-10 (δ_C 127.4), and the 18-CH₃ methyl protons (δ_H 2.41) showed cross-peaks with C-3 (δ_C 156.9) and C-5 (δ_C 130.7), which confirm the hydroxyl group is adjacent to this methyl group. The methyl (18-CH₃) is bound to C-4, and long-range correlation ($^4J_{HC}$) with carbonyl (C-6) was not observed. The presence of a carbonyl at C-6 can be confirmed by the high-resolution mass spectrometry (molecular formula: $C_{17}H_{14}O_4$) and NOE experiments.

Additionally, the common correlations observed in the HMBC spectrum of H-1, H-11, H-14 with C-9 (δ_C 137.5); H-2 and H-11 with C-10 (δ_C 127.4); H-1 and the 18-CH₃ methyl hydrogens with C-5 (δ_C 130.7); as well as H-11, H-14, the 15-CH₃ methyl hydrogens, and the 12-OCH₃ methoxyl hydrogens with C-12 (δ_C 163.8) confirm their proximity.

To confirm the assignment of the arrangement of the functional groups in the proposed structure, the nuclear Overhauser effects were analyzed after the straightforward process of decoupling the protons and acquiring the spectrum (Table 1). The selective irradiation of the resonance frequency of H-14 (δ_H 7.69) generated an NOE for the signal of 15-CH₃ at δ_H 2.16. Additionally, selective irradiation of the resonance frequency of H-11 (δ_H 7.47) caused an enhancement of the signals at δ_H 8.03 and δ_H 4.01 (H-1 and 12-OCH₃, respectively), which indicated that H-11 is close in space to the methoxyl group and to H-1. When H-1 (δ_H 8.03) was irradiated, an NOE was also observed at δ_H 7.47, confirming the proximity of H-1 and H-11. Selective irradiation of the resonance

frequency of H-2 (δ_{H} 7.17) intensified a single signal at δ_{H} 8.03, which corresponds to the signal of H-1. These results allowed us to propose a structure for this degraded diterpenoid.

Byrsonimaquinone can be considered a terpenoid derived from *ortho*-naphthoquinone. Naphthoquinones can be formed either from acetate/malonate or shikimate/2-oxoglutarate/isoprenoid pathways. Naphthoquinones, their derivatives, xanthenes, and terpene derivatives have been isolated previously from members of this genus.^{4,19,20,23} Quinones are derived from the oxidation of phenolic compounds; the oxidation of catechols (1,2-dihydroxybenzenes) leads to *ortho*-quinones, and the oxidation of quinols (1,4-dihydroxybenzenes) leads to *para*-quinones. Accordingly, terpenoid quinones are built up from shikimate and terpenoid pathways. Furthermore, a terpenoid quinone biosynthetic pathway is suggested for this compound, a combined metabolic pathway of shikimate-derived (quinone derivatives) and terpenoid pathways (diterpenoid moiety).³¹

All the compounds isolated from the roots of the second specimen were identified through comparison of their spectral data with the literature.²⁶⁻³⁰

CONCLUSION

This work contributed to our knowledge of the chemistry of the genus *Byrsonima*. Of the compounds isolated from the roots, the triterpenoids α -amyrin, β -amyrin, and oleanolic acid, and the mixture of stigmaterol, β -sitosterol and campesterol have been described before for other species, most of which have been isolated from *B. crassifolia*¹¹⁻¹⁴ and *B. verbascifolia*,^{21,22} except for glochidonol which has not been described in the genus so far.

The phytochemical investigation of the leaves and stems afforded the flavonoids (+)-catechin, (-)-epicatechin, quercitrin and isoquercitrin, which are frequently found in *Byrsonima* species including *B. basiloba*,^{6,7} *B. bucidaefolia*,⁸ *B. crassa*,^{9,10} *B. crassifolia*,¹¹⁻¹⁴ *B. fagifolia*,¹⁵⁻¹⁷ *B. intermedia*,¹⁸ *B. microphylla*,^{19,20} and *B. verbascifolia*.^{21,22} The investigation also afforded the tannins 3,5-di-*O*-galloylquinic acid, 5-*O*-galloylquinic acid, 5-*O*-(3-methylgalloyl)-quinic acid, 3,4,5-tri-*O*-galloylquinic acid and gallic acid, which have been previously described in *B. crassa*^{9,10} and *B. fagifolia*.¹⁵⁻¹⁷ Among the compounds isolated from the stems of *B. coccolobifolia*, (+)-syringaresinol and trigonostemone are new in the genus *Byrsonima*. Additionally, the degraded diterpenoid byrsonimaquinone, obtained from stems of *B. coccolobifolia*, is described for the first time in the literature.

SUPPLEMENTARY MATERIAL

The high-resolution mass spectrometry, TOF - Q II - ESI - TOF, ¹H NMR, HSQC, HMBC and 1D NOE spectra for isolated compounds (Figures 1S-14S) are freely available at <http://quimicanova.sbq.org.br> in PDF file.

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REFERENCES

- Guilhon-Simplicio, F.; Pereira, M. M.; *Quim. Nova* **2011**, *34*, 1032.
- Françoso, R. D.; Brandão, R.; Nogueirac, C. C.; Salmona, Y. B.; Machado, R. B.; Colli, G. R.; *Natureza & Conservação* **2015**, *13*, 35.
- Sannomiya, M.; Fonseca, V. B.; da Silva, M. A.; Rocha, L. R. M.; dos Santos, L. C.; Hiruma-Lima, C. A.; Souza Brito, A. R. M.; Vilegas, W.; *J. Ethnopharmacol.* **2005**, *97*, 1.
- de Sousa, L. R. F.; Ramalho, S. D.; Burger, M. C. M.; Nebo, L.; Fernandes, J. B.; da Silva, M. F. G. F.; Iemma, M. R. C.; Corrêa, C. J.; de Souza, D. H. F.; Lima, M. I. S.; Vieira, P. C.; *J. Nat. Prod.* **2014**, *77*, 392.
- de Sousa, L. R. F.; Ramalho, S. D.; Fernandes, J. B.; da Silva, M. F. G. F.; Iemma, M. R. C.; Corrêa, C. J.; de Souza, D. H. F.; Lima, M. I. S.; Vieira, P. C.; *J. Braz. Chem. Soc.* **2014**, *25*, 1832.
- Lira, W. M.; Santos, F. V.; Sannomiya, M.; Rodrigues, C. M.; Vilegas, W.; Varanda, E. A.; *J. Med. Food* **2008**, *11*, 111.
- Figueiredo, M. E.; Michelin, D. C.; Sannomiya, M.; Silva, M. A.; Santos, L. C.; Almeida, L. F. R.; Salgado, H. R. N.; Vilegas, W.; *Rev. Bras. Cienc. Farm.* **2005**, *41*, 79.
- Castillo-Avila, G. M.; García-Sosa, K.; Peña-Rodríguez, L. M.; *Nat. Prod. Commun.* **2009**, *4*, 83.
- Sannomiya, M.; Rodrigues, C. M.; Coelho, R. G.; Santos, L. C.; Hiruma-Lima, C. A.; Brito, A. R. M. S.; Vilegas, W.; *J. Chromatogr. A* **2004**, *1035*, 47.
- Sannomiya, M.; Fonseca, V. B.; Silva, M. A.; Rocha, L. R.; Santos, L. C.; Hiruma-Lima, C. A.; Souza Brito, A. R.; Vilegas, W.; *J. Ethnopharmacol.* **2005**, *97*, 1.
- Geiss, F.; Heinrich, M.; Hunkler, D.; Rimpler, H.; *Phytochemistry* **1995**, *39*, 635.
- Rastrelli, L.; De Tommasi, N.; Berger, I.; Caceres, A.; Saravia, A.; De Simone, F.; *Phytochemistry* **1997**, *45*, 647.
- Amarquaye, A.; Che, C.; Bejar, E.; Malone, M. H.; Fong, H. H. S.; *Planta Med.* **1994**, *60*, 85.
- Bejar, E.; Amarquaye, A.; Che, C.; Malone, M. H.; Fong, H. H. S.; *Int. J. Pharmacogn.* **1995**, *33*, 25.
- Lima, Z. P.; Santos, R. C.; Torres, T. U.; Sannomiya, M.; Rodrigues, C. M.; Santos, L. C.; Pellizzon, C. H.; Rocha, L. R. M.; Vilegas, W.; Brito, A. R. M. S.; Cardoso, C. R. P.; Varanda, E. A.; Moraes, H. P.; Bauab, T. M.; Carli, C.; Carlos, I. Z.; Hiruma Lima, C. A.; *J. Ethnopharmacol.* **2008**, *120*, 149.
- Higuchi, C. T.; Sannomiya, M.; Pavan, F. R.; Leite, S. R. A.; Sato, D. N.; Franzblau, S. G.; Sacramento, L. V. S.; Vilegas, W.; Leite, C. Q. F.; *J. Evidence-Based Complementary Altern. Med.* **2011**, ID 128349.
- Sannomiya, M.; Santos, L. C.; Carbone, V.; Napolitano, A.; Piacente, S.; Pizza, C.; Souza-Brito, A. R. M.; Vilegas, W.; *Rapid Commun. Mass Spectrom.* **2007**, *21*, 1393.
- Sannomiya, M.; Cardoso, C. R. P.; Figueiredo, M. E.; Rodrigues, C. M.; Santos, L. C.; Santos, F. V.; Serpeloni, J. M.; Colus, I. M.; Vilegas, W.; Varanda, E. A.; *J. Ethnopharmacol.* **2007**, *112*, 319.
- Aguiar, R. M.; David, J. P.; David, J. M.; *Phytochemistry* **2005**, *66*, 2388.
- Rocha, J. H.; Cardoso, M. P.; David, J. P.; David, J. M.; *Biosci., Biotechnol., Biochem.* **2006**, *70*, 2759.
- Gottlieb, O. R.; Mendes, P. H.; Magalhães, M. T.; *Phytochemistry* **1975**, *14*, 1456.
- Dosse, C.; Morreti, C.; Tessier, A. M.; Delaveau, P.; *Plant. Med. Phytother.* **1980**, *14*, 136.
- Lorenzi, K. C.; Rodrigues, C. M.; Sannomiya, M.; de Almeida, L. F. R.; Brito, A. R. M. S.; Vilegas, W.; *Resumos da 29ª Reunião Anual da Sociedade Brasileira de Química, Águas de Lindóia, Brasil, 2006*.
- Lorenzi, K. C.; Rodrigues, C. M.; Sannomiya, M.; Rinaldo, D.; Brito, A. R. M. S.; Vilegas, W.; *Resumos da 30ª Reunião Anual da Sociedade Brasileira de Química, Águas de Lindóia, Brasil, 2007*.

25. Tene, M.; Tane, P.; Tamokou, J. de D.; Kuate, J. R.; Cnnolly, J. D.; *Phytochem. Lett.* **2008**, *1*, 120.
26. Vieira-Junior, G. M.; Souza, C. M. L.; Chaves, M. H.; *Quim. Nova* **2005**, *28*, 183.
27. Seebacher, W.; Simic, N.; Weis, R.; Saf, R.; Kunert, O.; *Magn. Reson. Chem.* **2003**, *41*, 636.
28. Puapairoj, P.; Naengchomng, W.; Kijjoa, A.; Pinto, M. M.; Pedro, M.; Nascimento, M. S. J.; Silva, A. M. S.; Herz, W.; *Planta Med.* **2005**, *71*, 208.
29. Garg, V. K.; Nes, W. R.; *Phytochemistry* **1984**, *23*, 2925.
30. Costa, E. V.; Sampaio, M. F. C.; Salvador, M. J.; Nepel, A.; Barison, A.; *Quim. Nova* **2015**, *38*, 769.
31. Dewick, P. M.; *Medicinal Natural Products: A Biosynthetic Approach*, 2nd ed., John Wiley & Sons, Ltd: New York, 2002.