

TERNARY SURFACE COMPLEX: COADSORPTION OF Cu(II), Zn(II), Cd(II) AND NITRILOTRIS(METHYLENE PHOSPHONIC) ACID ONTO BOEHMITE

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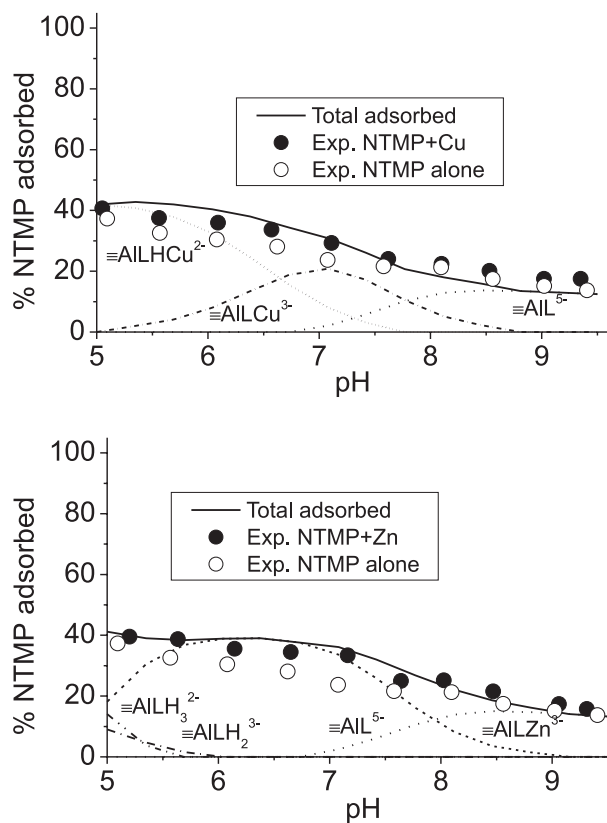


Figura 1S. Percentage NTMP 5×10^{-4} M adsorbed vs. pH in the presence of Cu(II) and Zn(II) 5×10^{-4} M. The curves are calculated using the constants from Table 1. Conditions: 1 g/L boehmite, 0.1 M NaNO_3

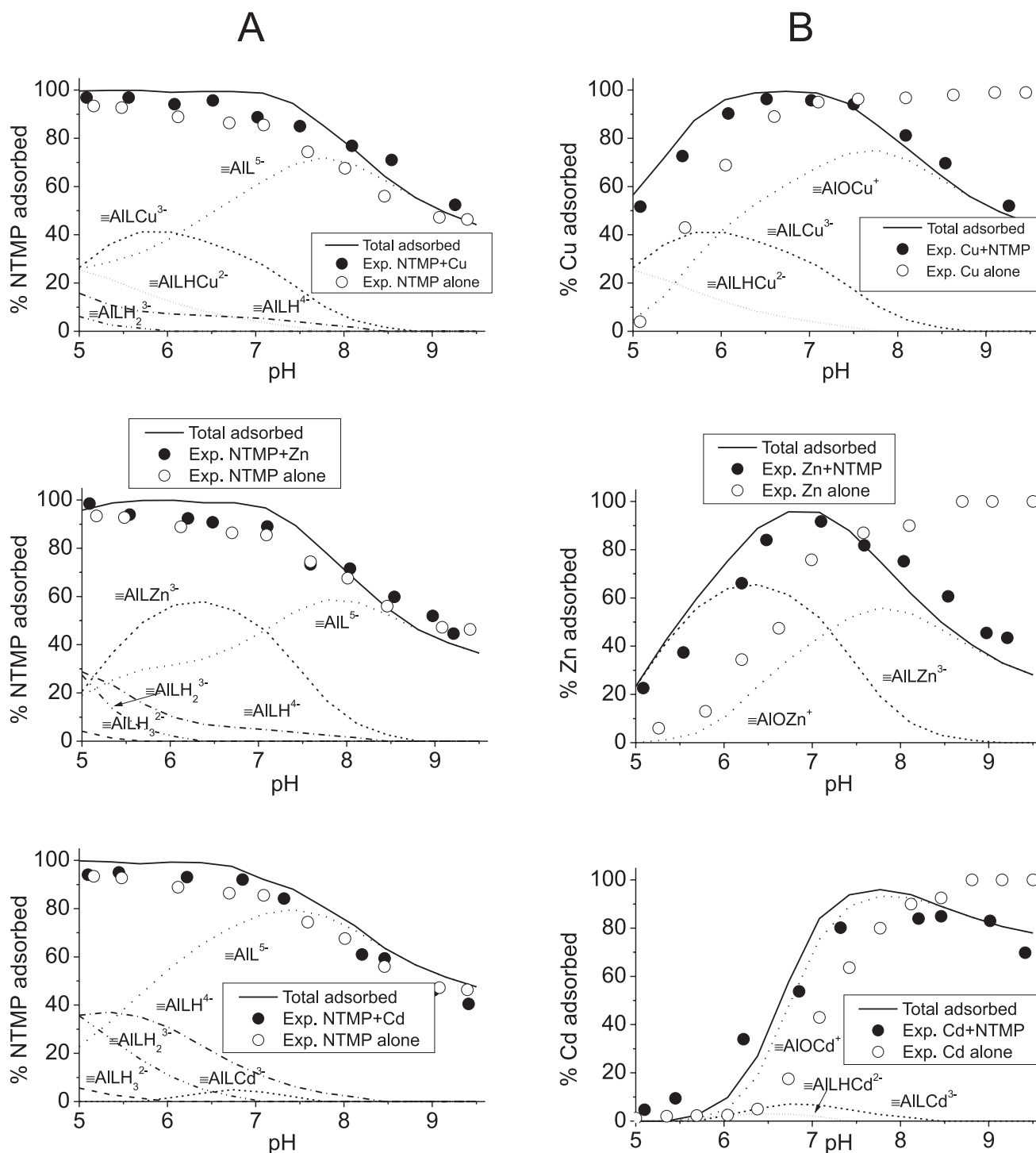


Figura 2S. A. Percentage NTMP 1×10^{-4} M adsorbed vs. pH in the presence of Cu(II), Zn(II) and Cd(II) 1×10^{-4} M. The curves are calculated using the constants from Table 1. Conditions: 1 g/L boehmite, 0.1 M NaNO_3 . B. Percentage Cu(II), Zn(II) and Cd(II) 1×10^{-4} M adsorbed vs. pH in the presence of NTMP 1×10^{-4} M. The curves are calculated using the constants from Table 1. Conditions: 1 g/L boehmite, 0.1 M NaNO_3 .

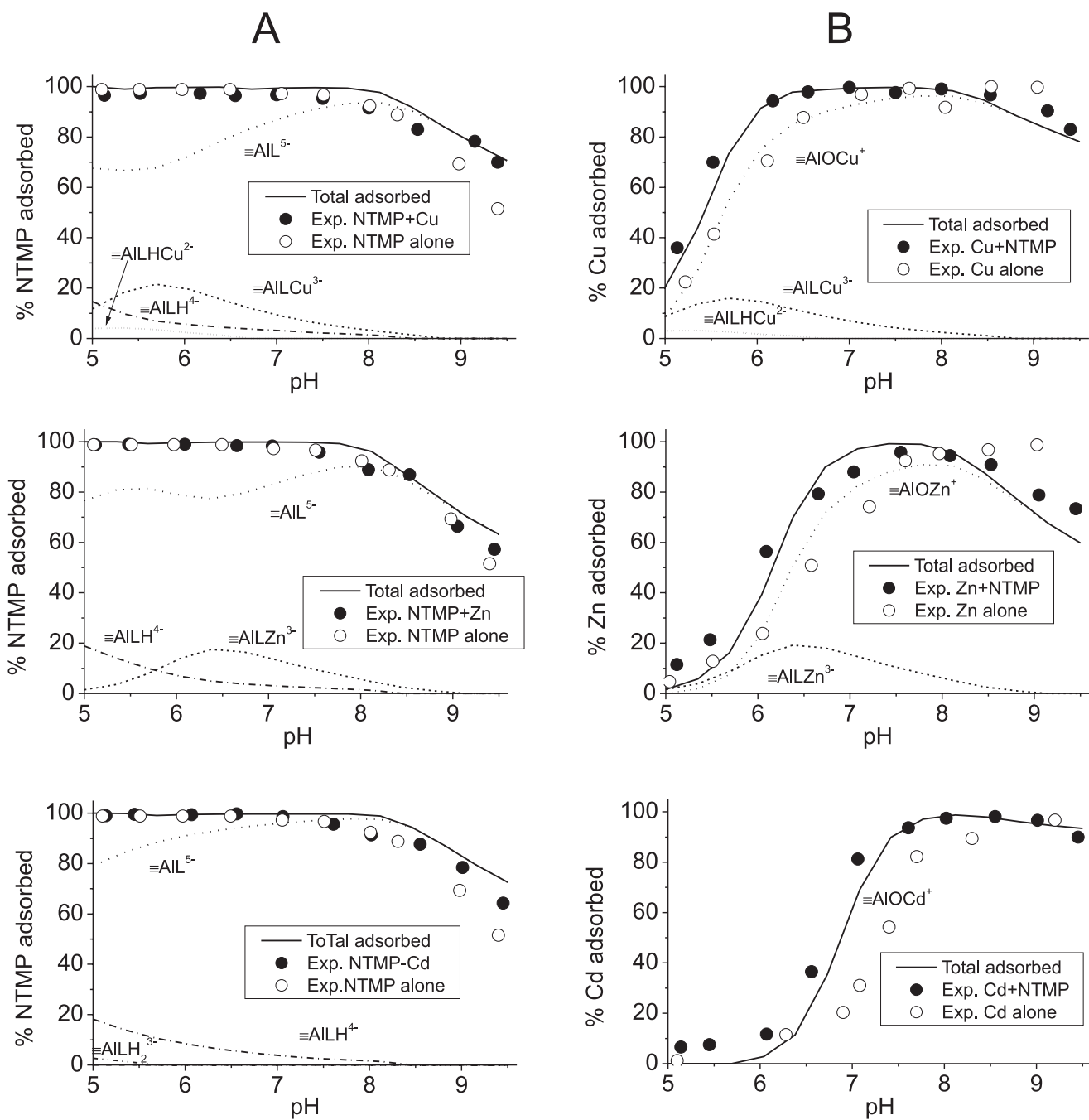


Figure 3S. A. Percentage NTMP 5×10^{-5} M adsorbed vs. pH in the presence of Cu(II), Zn(II) and Cd(II) 5×10^{-5} M. The curves are calculated using the constants from Table 1. Conditions: 1 g/L boehmite, 0.1 M NaNO_3 . B. Percentage Cu(II), Zn(II) and Cd(II) 5×10^{-5} M adsorbed vs. pH in the presence of NTMP 5×10^{-5} M. The curves are calculated using the constants from Table 1. Conditions: 1 g/L boehmite, 0.1 M NaNO_3 .

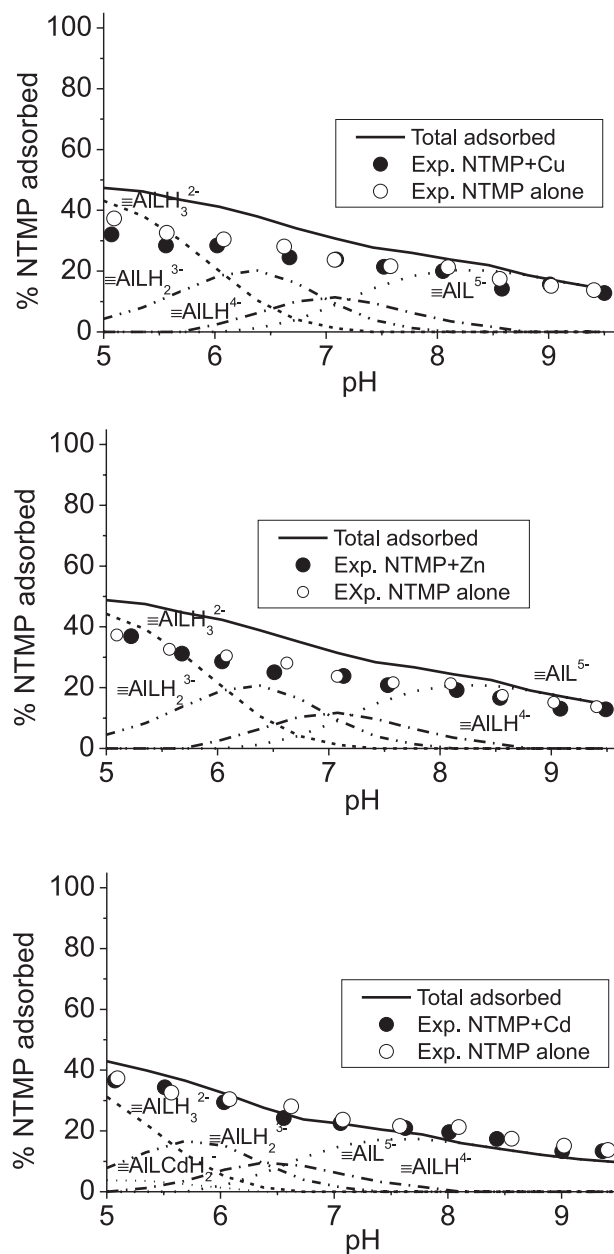


Figura 4S. Percentage NTMP 5×10^{-4} M adsorbed vs. pH in the presence of Cu(II), Zn(II) and Cd(II) 1×10^{-4} M. The curves are calculated using the constants from Table 1. Conditions: 1 g/L boehmite, 0.1 M NaNO_3

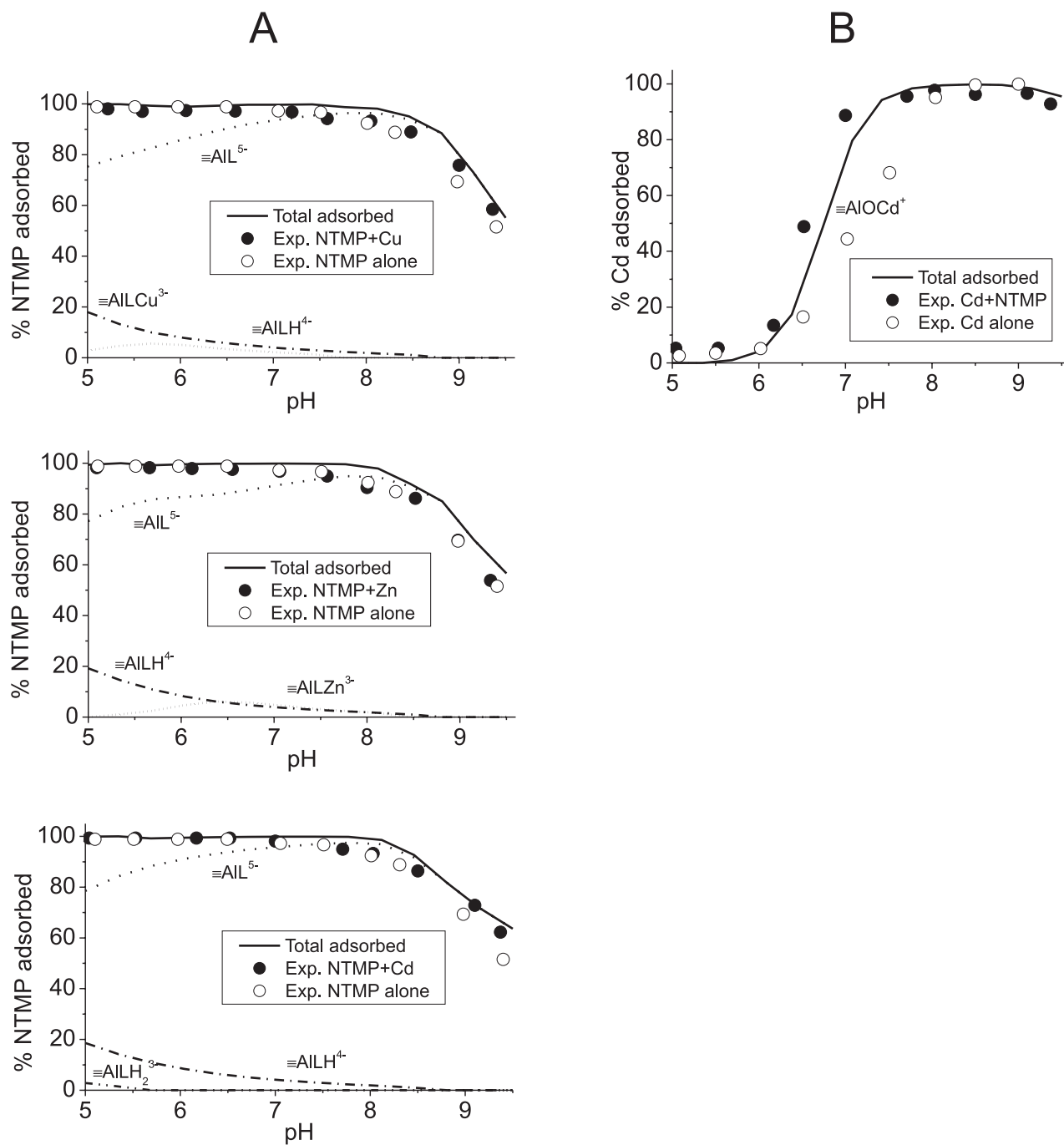


Figure 5S. A. Percentage NTMP 5×10^{-5} M adsorbed vs. pH in the presence of Cu(II), Zn(II) and Cd(II) 1×10^{-5} M. The curves are calculated using the constants from Table 1. Conditions: 1 g/L boehmite, 0.1 M NaNO_3 . B. Percentage Cd(II) 1×10^{-5} M adsorbed vs. pH in the presence of NTMP 5×10^{-5} M. The curves are calculated using the constants from Table 1. Conditions: 1 g/L boehmite, 0.1 M NaNO_3 .